



Article

Green Synthesis of High Temperature Stable Anatase Titanium Dioxide Nanoparticles Using Gum Kondagogu: Characterization and Solar Driven Photocatalytic Degradation of Organic Dye

Saranya K. S. -Kothaplamoottil Sivan Saranya ¹, Vinod Vellora Thekkae Padil ^{2,*}, Chandra Senan ³, Rajendra Pilankatta ⁴, Saranya K. -Kunjumon Saranya ¹, Bini George ^{1,*}, Stanisław Wacławek ² and Miroslav Černík ^{2,*}

- Department of Chemistry, School of Physical Sciences, Central University of Kerala, Kerala 671316, India; sharanyacks@gmail.com (S.K.S.-K.S.S.); kunjumonsaranya916@gmail.com (S.K.-K.S.)
- Institute for Nanomaterials, Advanced Technologies and Innovation (CXI), Technical University of Liberec (TUL), Studentská 1402/2, 46117 Liberec 1, Czech Republic; stanislaw.waclawek@tul.cz
- Centre for Water Soluble Polymers, Applied Science, Faculty of Arts, Science and Technology, Wrexham Glyndwr University, Wrexham LL11 2AW, Wales, UK; c.senan@glyndwr.ac.uk
- Department of Biochemistry and Molecular Biology, School of Biological Sciences, Central, University of Kerala, Kerala 671316, India; praj74@gmail.com
- * Correspondence: vinod.padil@tul.cz (V.V.T.P.); binigeorgek@gmail.com (B.G.); miroslav.cernik@tul.cz (M.Č.); Tel.: +420-723-372-911 (V.V.T.P. & M.Č.)

Received: 28 October 2018; Accepted: 1 December 2018; Published: 4 December 2018



Abstract: The present study reports a green and sustainable method for the synthesis of titanium dioxide (TiO₂) nanoparticles (NPs) from titanium oxysulfate solution using Kondagogu gum (*Cochlospermum gossypium*), a carbohydrate polymer, as the NPs formation agent. The synthesized TiO₂ NPs were categorized by techniques such as X-Ray Diffraction (XRD), Fourier transform infrared (FTIR) spectroscopy analysis, Raman spectroscopy, scanning electron microscope- Energy-dispersive X-ray spectroscopy (SEM-EDX), Transmission electron microscopy (TEM), High-resolution transmission electron microscopy (HR-TEM), UV-visible spectroscopy, Brunauer-Emmett-Teller (BET) surface area and particle size analysis. Additionally, the photocatalytic actions of TiO₂ NPs were assessed with regard to their ability to degrade an organic dye (methylene blue) from aqueous solution in the presence of solar light. Various parameters affecting the photocatalytic activity of the TiO₂ NPs were examined, including catalyst loading, reaction time, pH value and calcination temperature of the aforementioned particles. This green synthesis method involving TiO₂ NPs explores the advantages of inexpensive and non-toxic precursors, the TiO₂ NPs themselves exhibiting excellent photocatalytic activity against dye molecules.

Keywords: titanium dioxide nanoparticles; green synthesis; gum kondagogu; methylene blue; photocatalysis

1. Introduction

Nanoparticles (metal and metal oxides) of various types have been widely employed via physical and chemical methods. Although these systems have resulted in the formation of numerous extremely diverse nanostructures, many environmental toxicity issues have emerged [1]. Metal oxides have recently been widely explored because of the huge variety of structural, material and functional properties exhibited by their nanoparticles. Transition metal oxide nanoparticles have generated